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Tris is a ,
buffer , that's
used to
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stable pH when
working with ,
solutions , in
the lab. In
this video,
we'll
demonstrate the

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shows how to
(correctly!)
make , buffers

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, and pH them.

Remember: 1)

Weigh dry

ingredients 2)

Add water up to

70-80% of

final ...

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expression used
to , calculate
, the pH of a ,
buffer solution
, followed by
three different
style

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a buffer
solution when
given molarity
and volume of
reactants

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a buffer
solution when
given molarity

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two main
situations
where you get a
, buffer , :
making a ,
buffer , for a
specific pH by

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combining a
weak acid and a
salt ...

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